

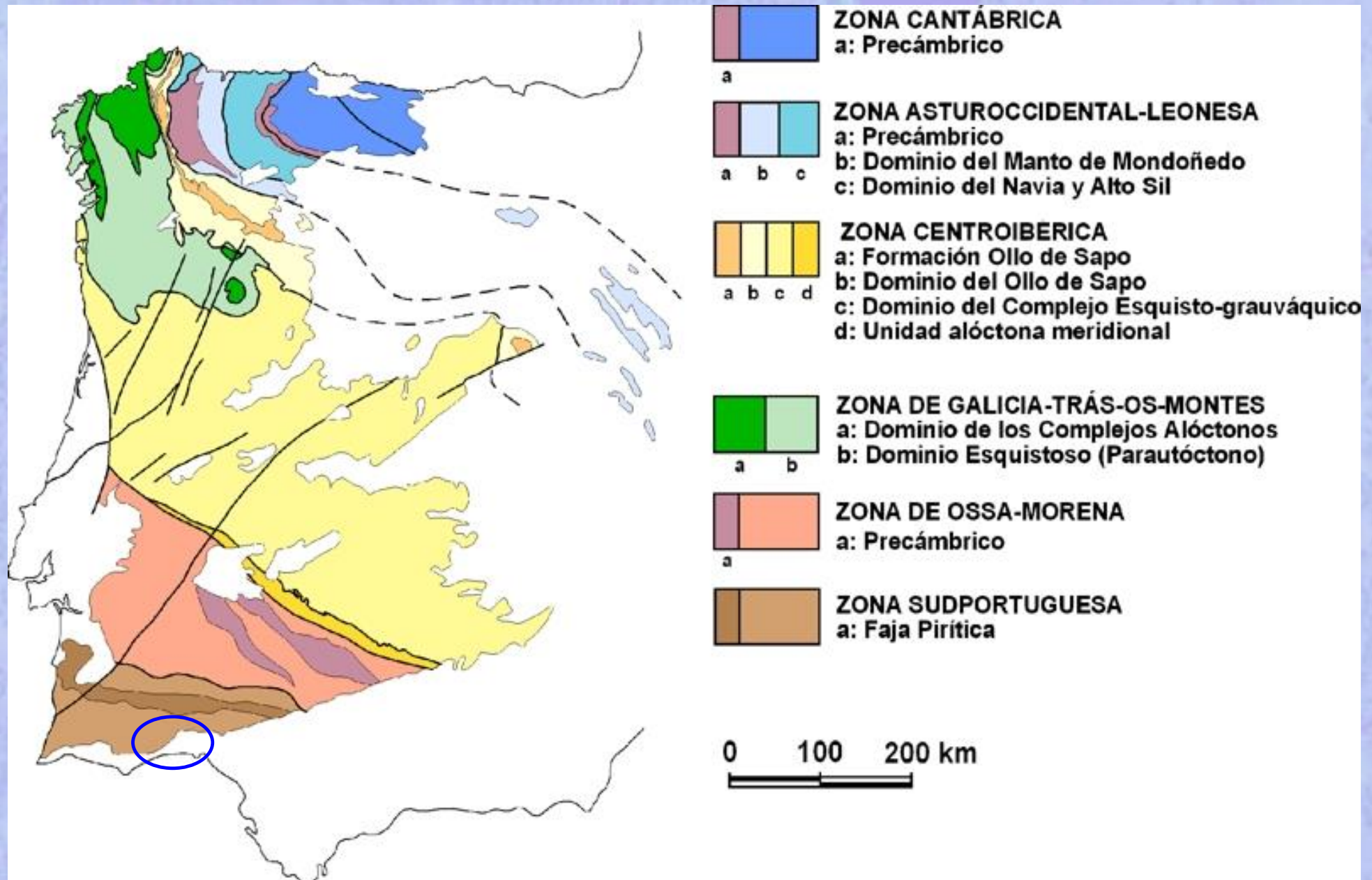
IRON SULPHIDE MINEWASTE CONTAMINATION MAPPING USING HYPERSENSITIVE DATA

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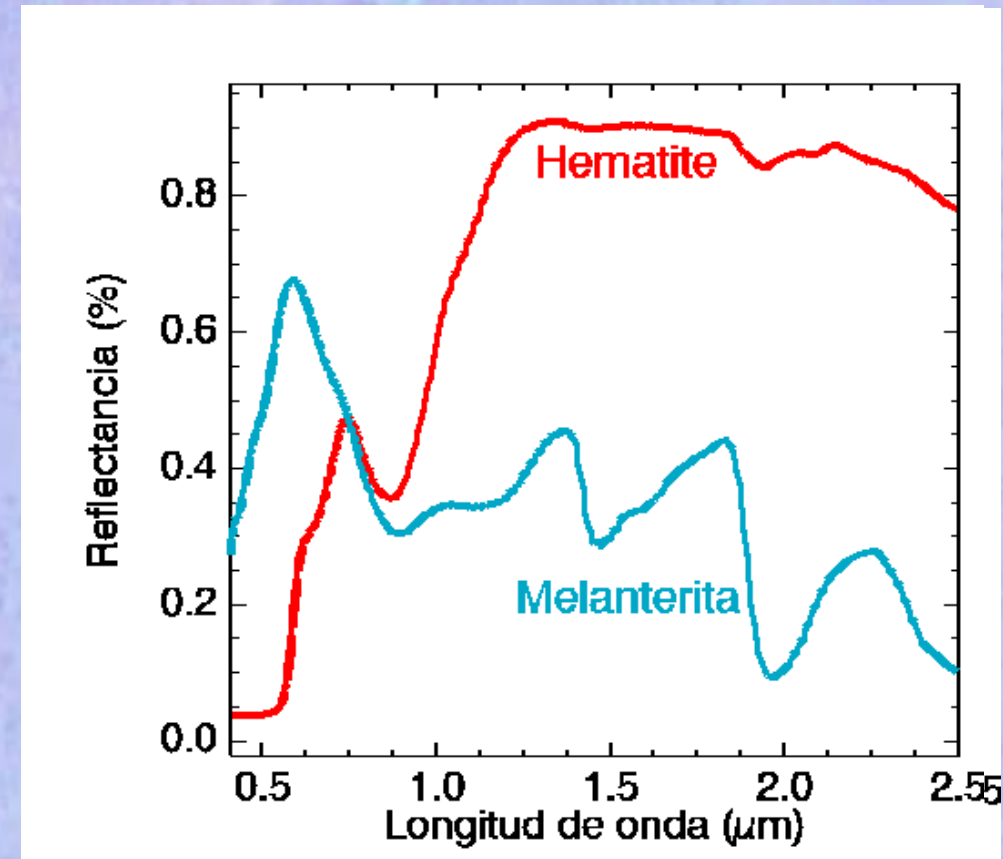
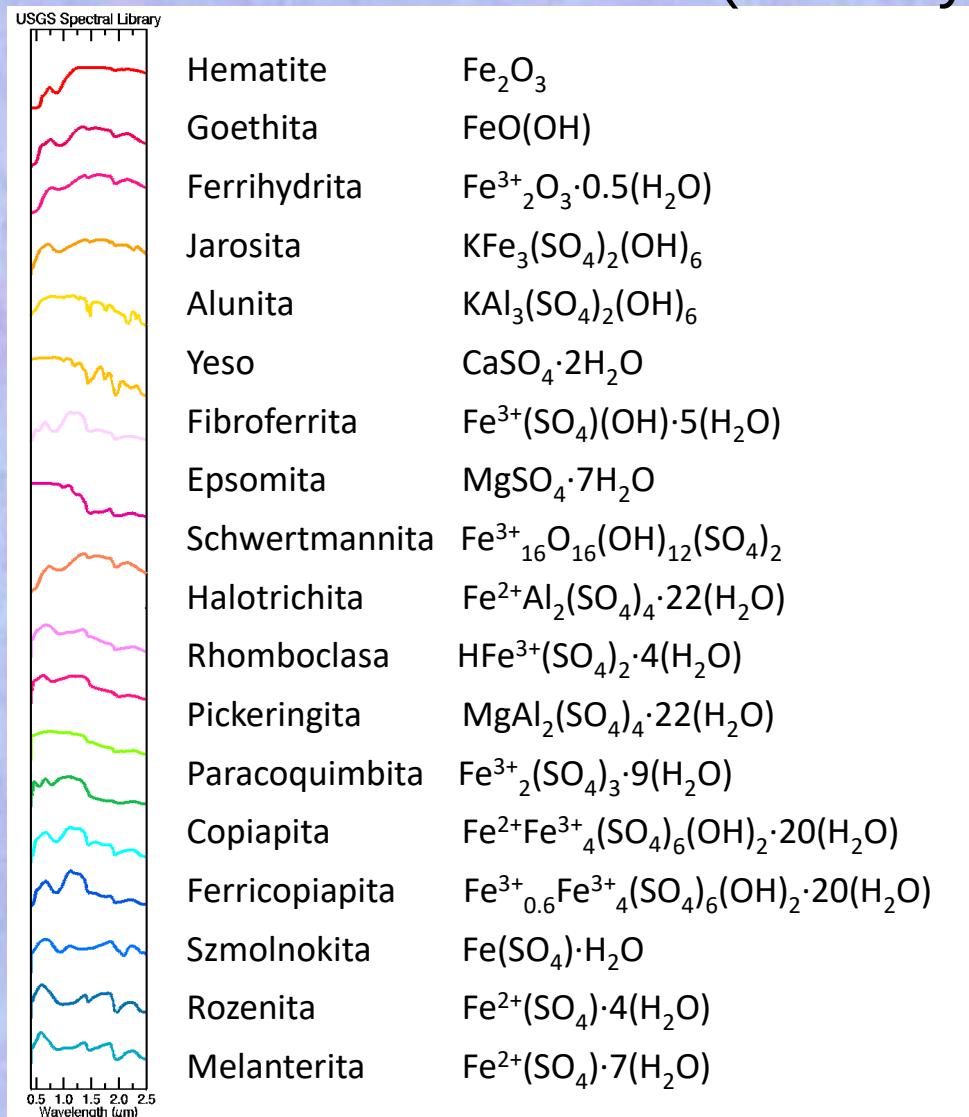
³Universidad de León, Facultad de Ciencias Ambientales, Campus de Vegazana s/n, León, España. egarm@unileon.es



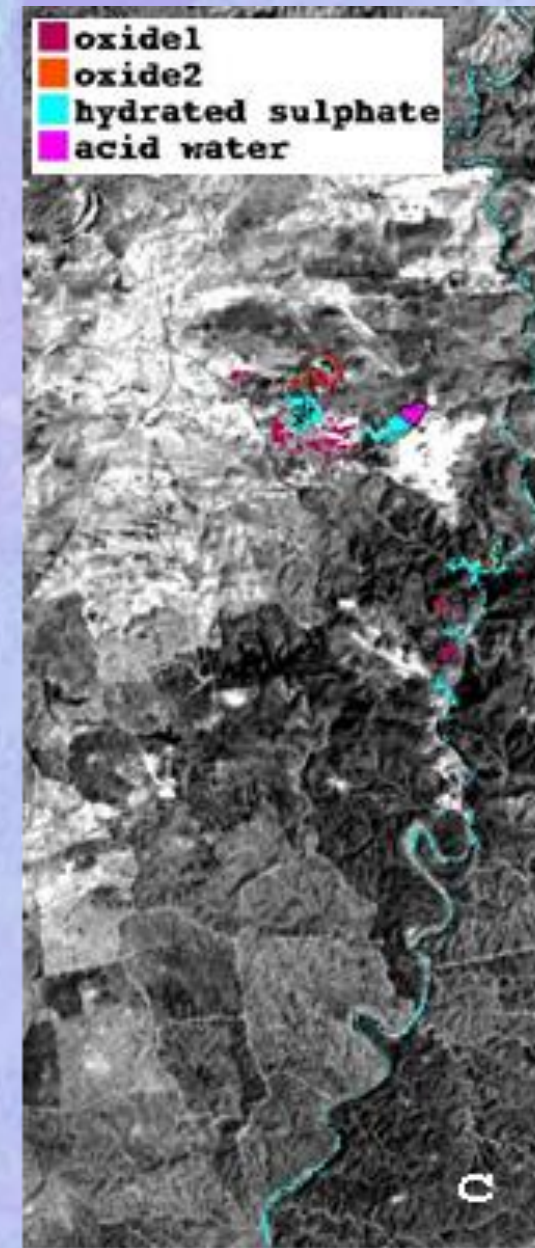
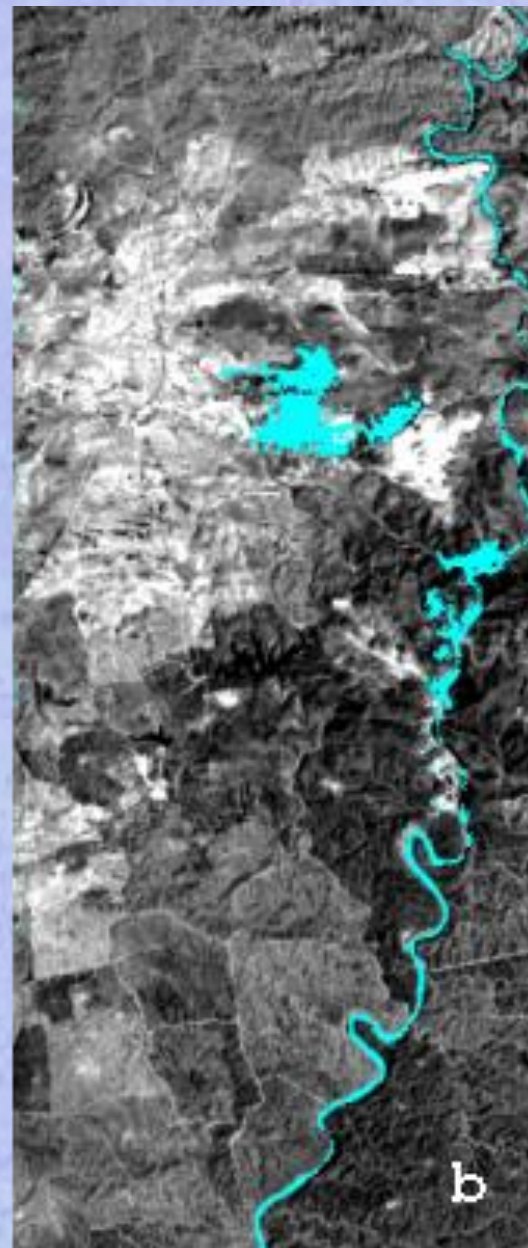
BUZZI, J., 2012, Imaging spectroscopy to evaluate the contamination from sulphide mine waste in the Iberian Pyrite Belt using hyperspectral sensors (Huelva, Spain), Tesis Doctoral Universidad de León, 212 p.

RIAZA, J. BUZZI, E. GARCÍA-MELÉNDEZ , V. CARRÈRE and A. MÜLLER, 2011, Monitoring the extent of contamination from acid mine drainage in the Iberian Pyrite Belt (SW Spain) using hyperspectral imagery, Remote Sensing, 2011, 3, 2166-2186; doi:10.3390/rs3102166.

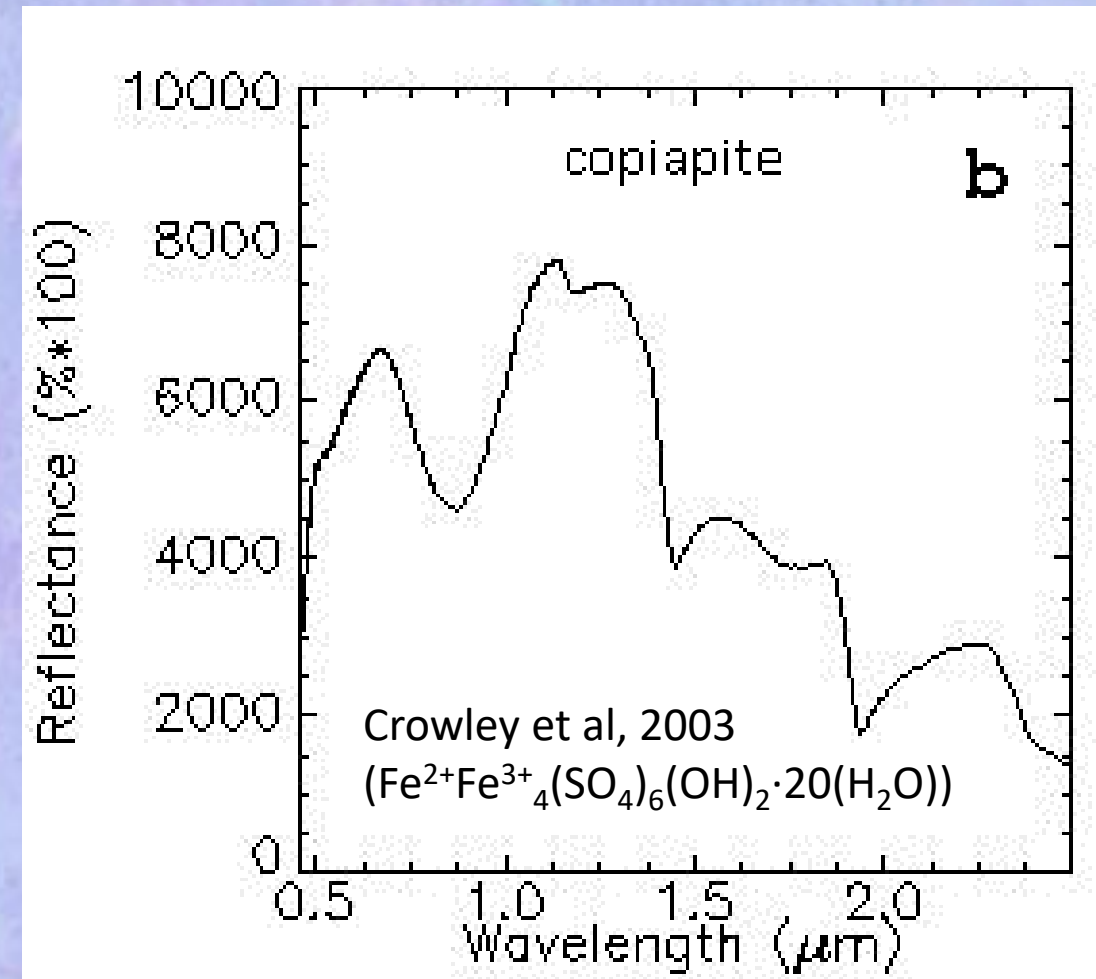
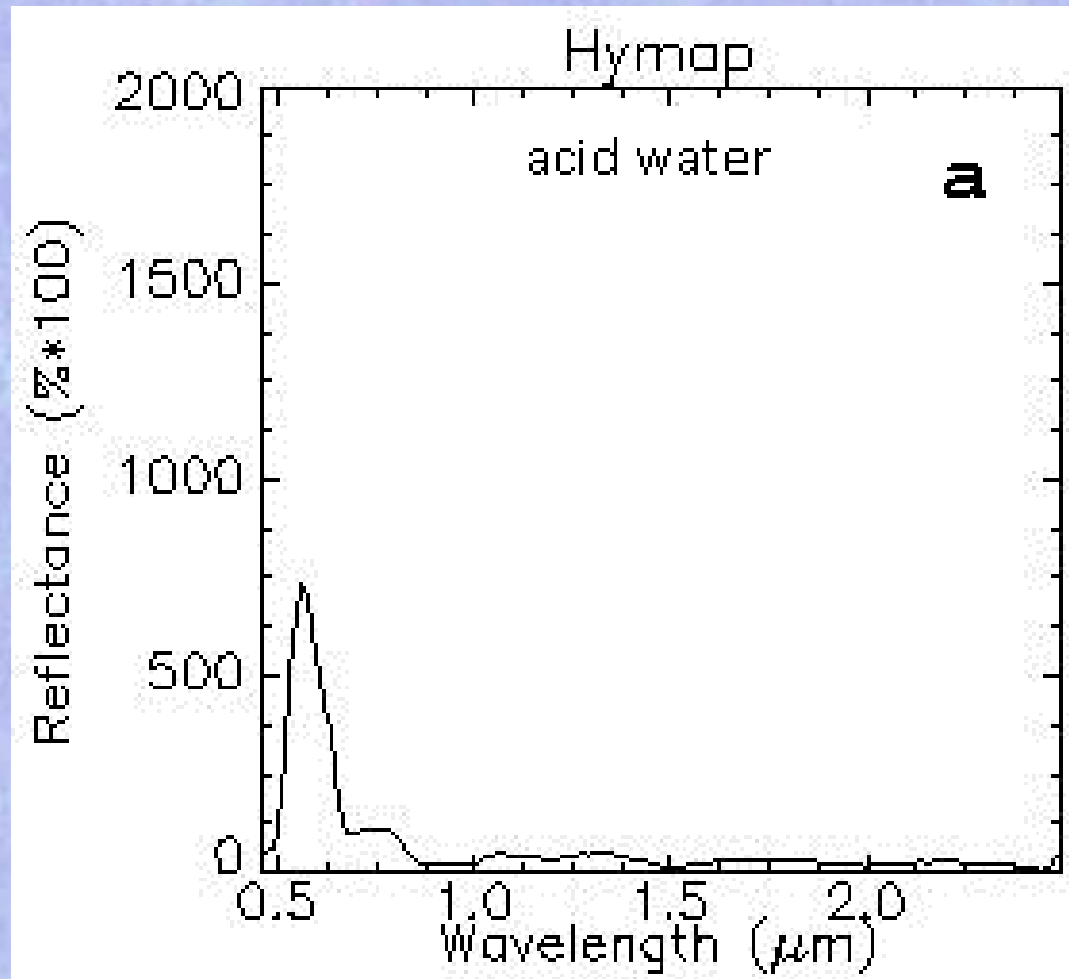
SPECTRAL LIBRARY OF MINERALS PRODUCT OF IRON SULPHIDE WEATHERING (Crowley et al, 2003)



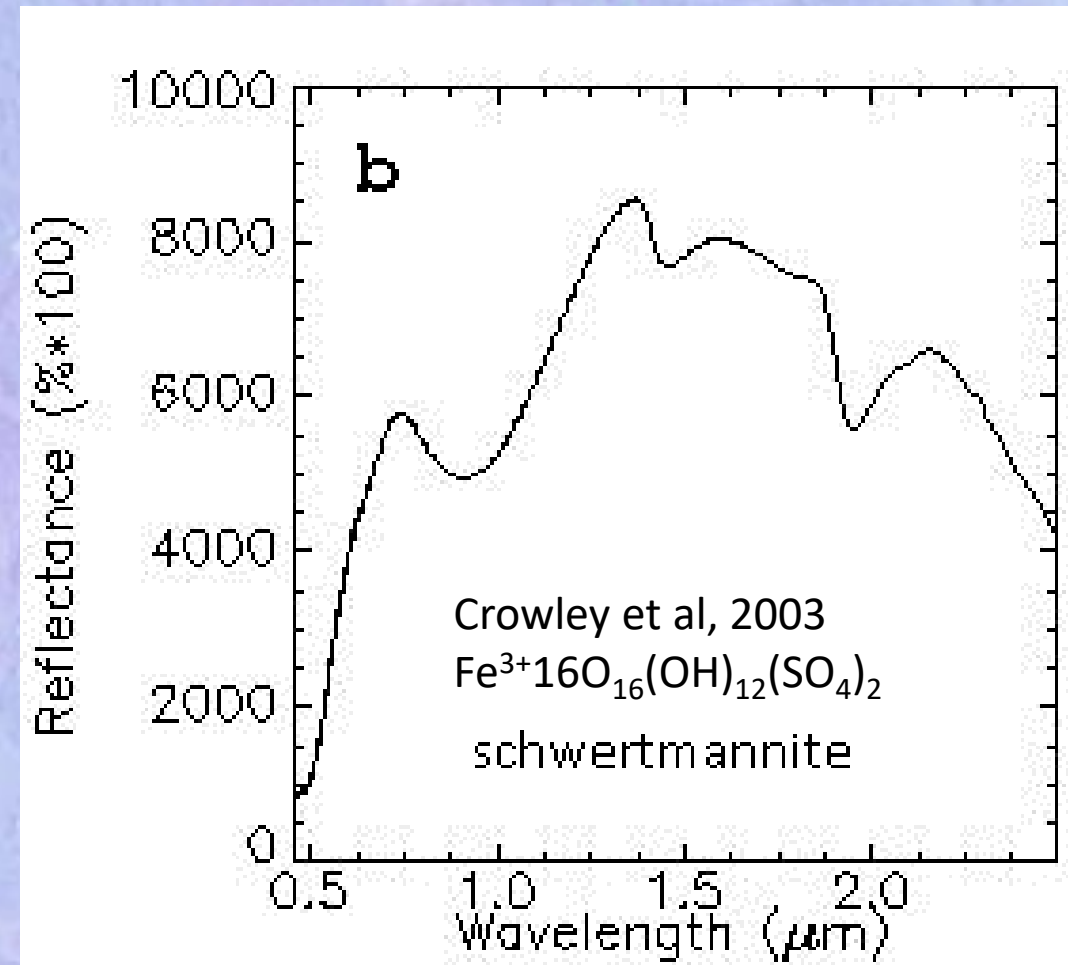
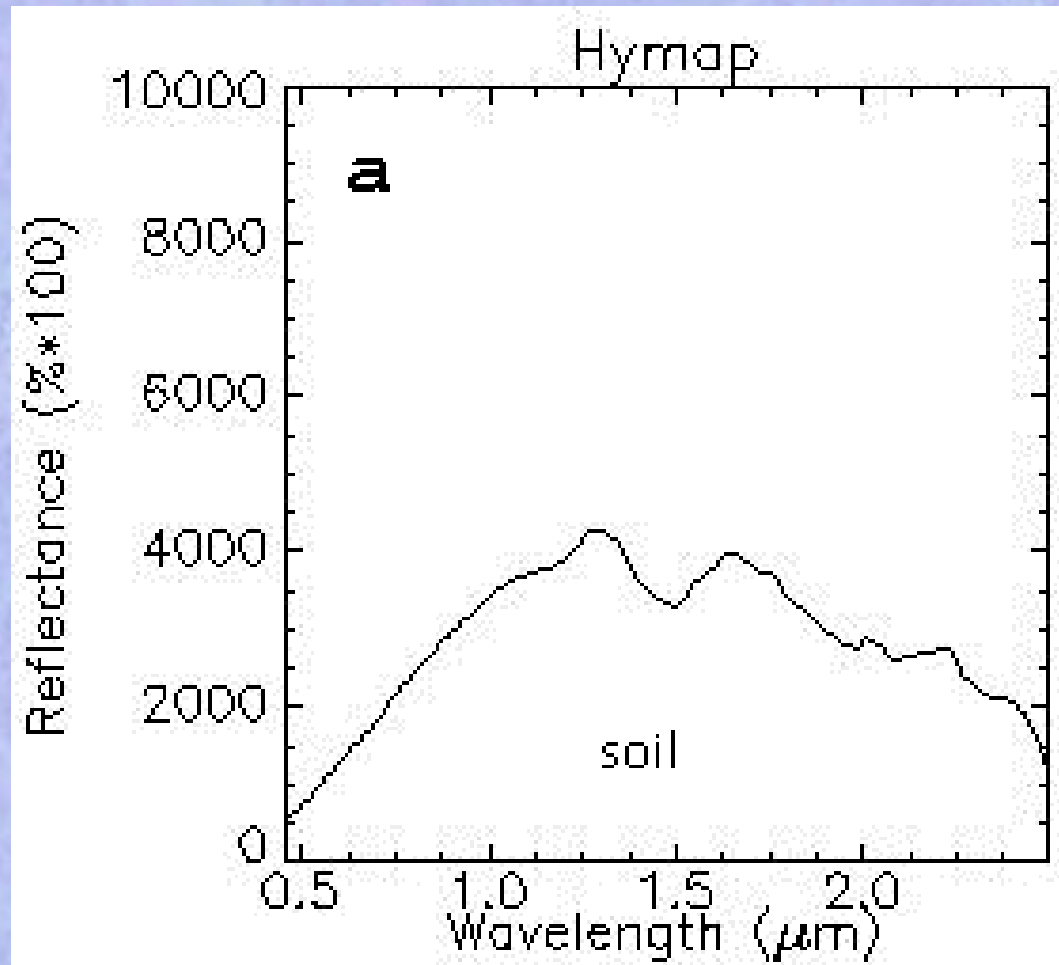
HYPERION 2005



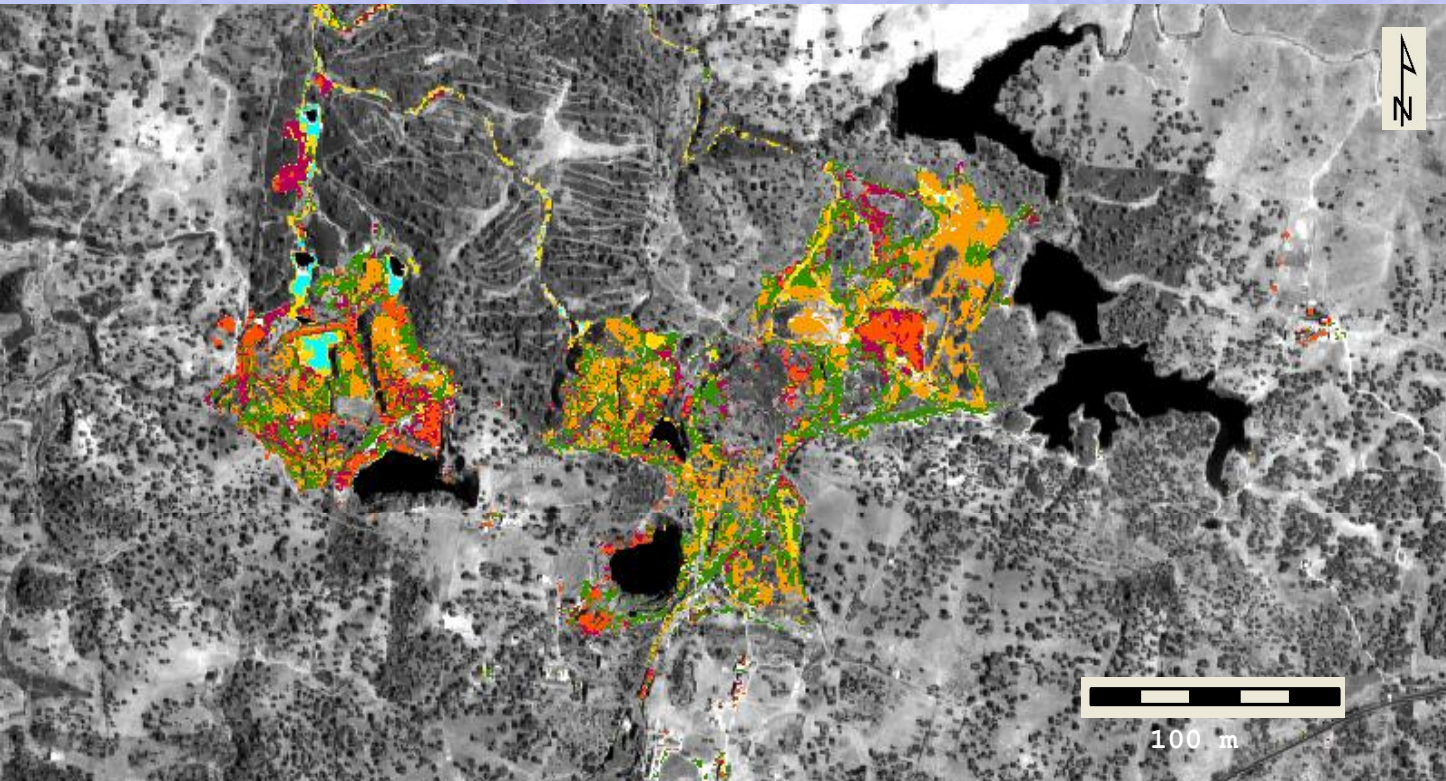
SPECTRAL ANALYST WITH REFERENCE SPECTRAL LIBRARIES



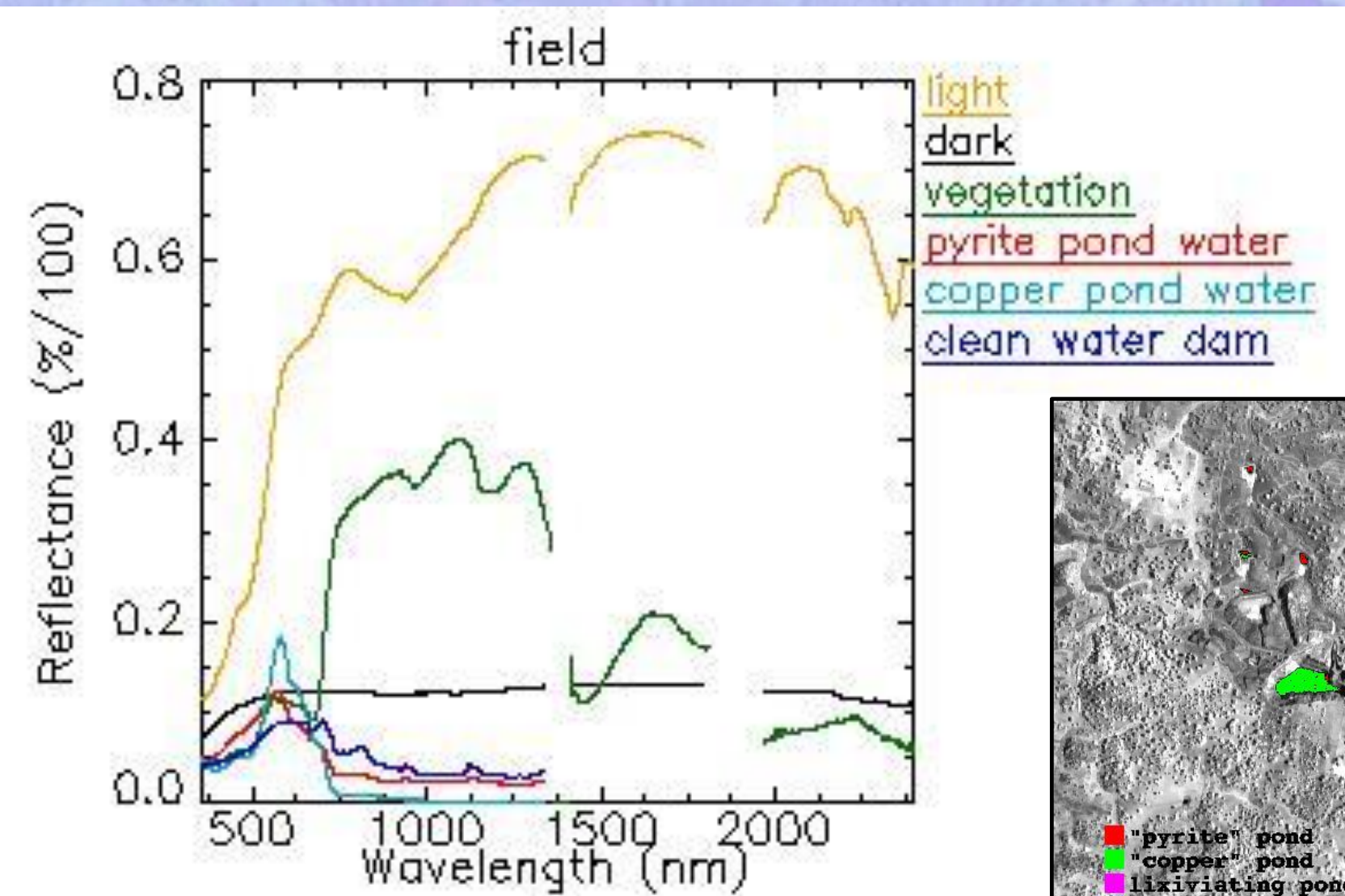
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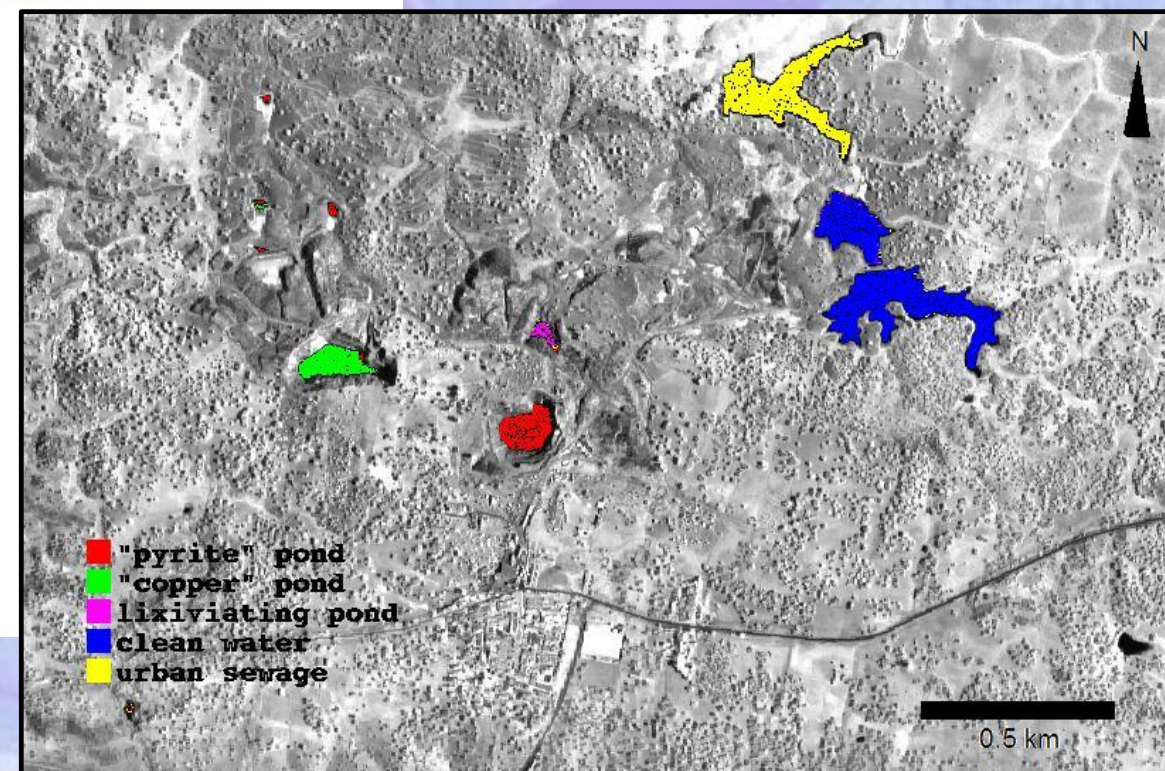
HYMAP August 2009



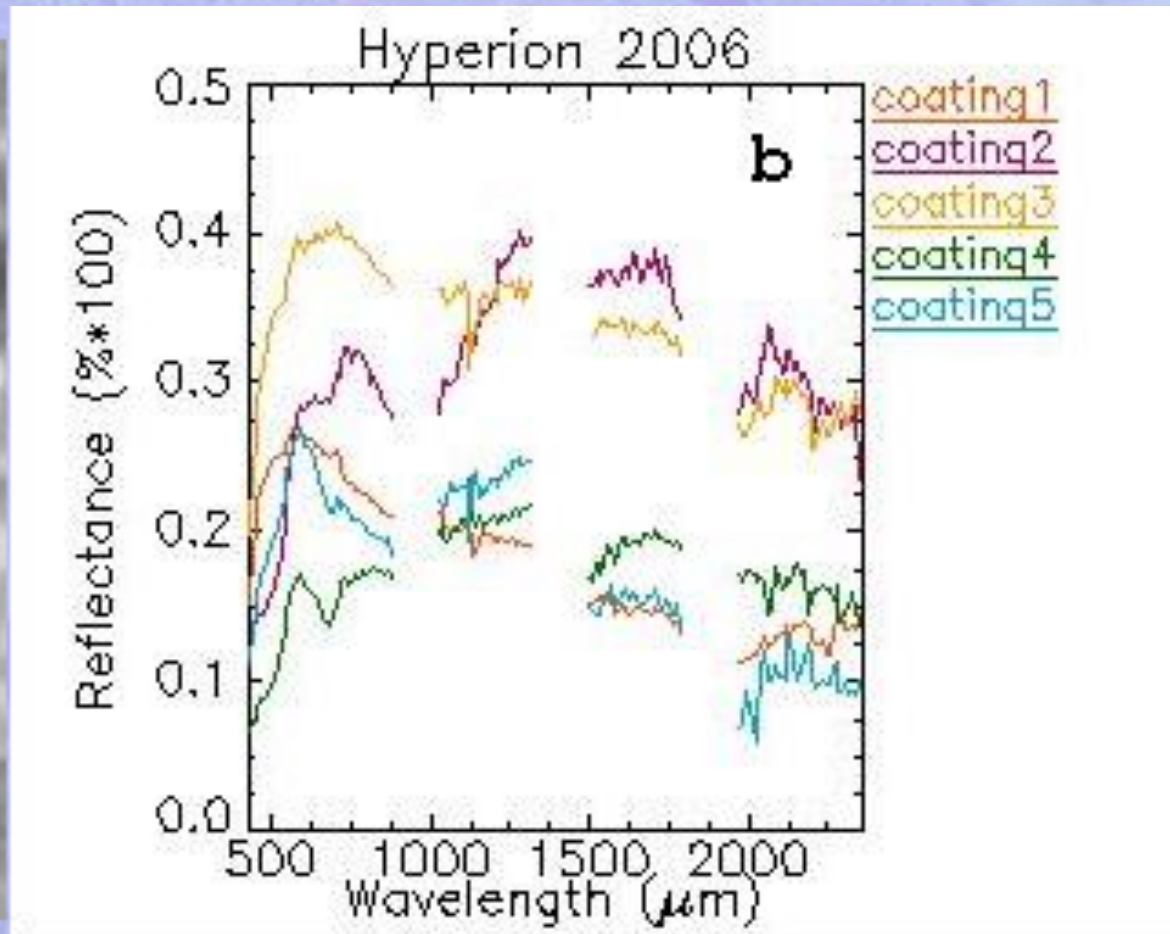
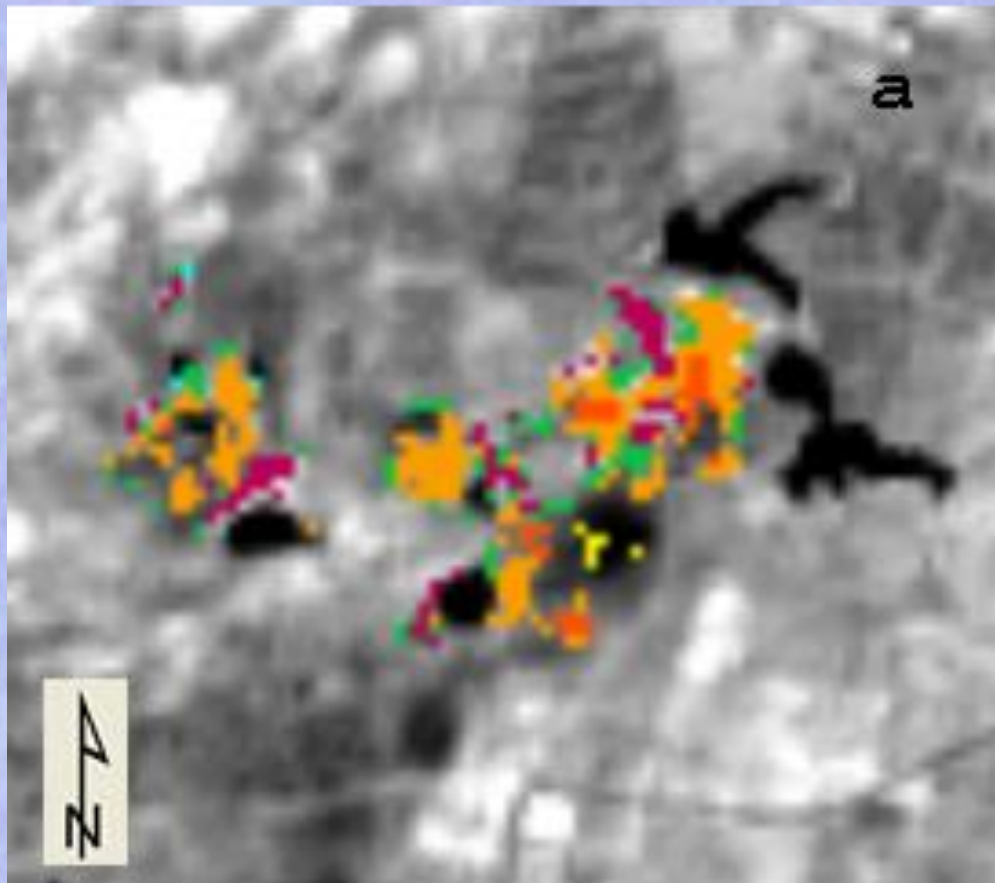
	Hematite	Fe_2O_3
	Goethite	$\text{FeO}(\text{OH})$
	Ferrihydrite	$\text{Fe}^{3+}_2\text{O}_3 \cdot 0.5(\text{H}_2\text{O})$
	Jarosite	$(\text{SO}_4)_2\text{KFe}_3(\text{OH})_6$
	Alunite	$\text{KAl}_3(\text{SO}_4)_2(\text{OH})_6$
	Gypsum	$\text{SO}_4\text{Ca} \cdot 2\text{H}_2\text{O}$
	Fibroferrite	$\text{Fe}^{3+}(\text{SO}_4)(\text{OH}) \cdot 5(\text{H}_2\text{O})$
	Epsomite	$\text{MgSO}_4 \cdot 7(\text{H}_2\text{O})$
	Schwertmannite	$\text{Fe}^{3+}_{16}\text{O}_{16}(\text{OH})_{12}(\text{SO}_4)_2$
	Halotrichite	$\text{Fe}^{2+}\text{Al}_2(\text{SO}_4)_4 \cdot 22(\text{H}_2\text{O})$
	Rhomboclase	$\text{HFe}^{3+}(\text{SO}_4)_2 \cdot 4(\text{H}_2\text{O})$
	Pickeringite	$\text{MgAl}_2(\text{SO}_4)_4 \cdot 22(\text{H}_2\text{O})$
	Paracoquimbite	$\text{Fe}^{3+}_2(\text{SO}_4)_3 \cdot 9(\text{H}_2\text{O})$
	Copiapite	$\text{Fe}^{3+}\text{Fe}^{3+}_4(\text{SO}_4)_6(\text{OH})_2 \cdot 20(\text{H}_2\text{O})$
	Ferricopiapite	$\text{Fe}^{3+}_{0.6666}\text{Fe}^{3+}_4(\text{SO}_4)_6(\text{OH})_2 \cdot 20(\text{H}_2\text{O})$
	Szmolnokite	$\text{Fe}(\text{SO}_4) \cdot \text{H}_2\text{O}$
	Rozenite	$\text{Fe}^{2+}(\text{SO}_4) \cdot 4(\text{H}_2\text{O})$
	Melanterite	$\text{Fe}^{2+}(\text{SO}_4) \cdot 7(\text{H}_2\text{O})$



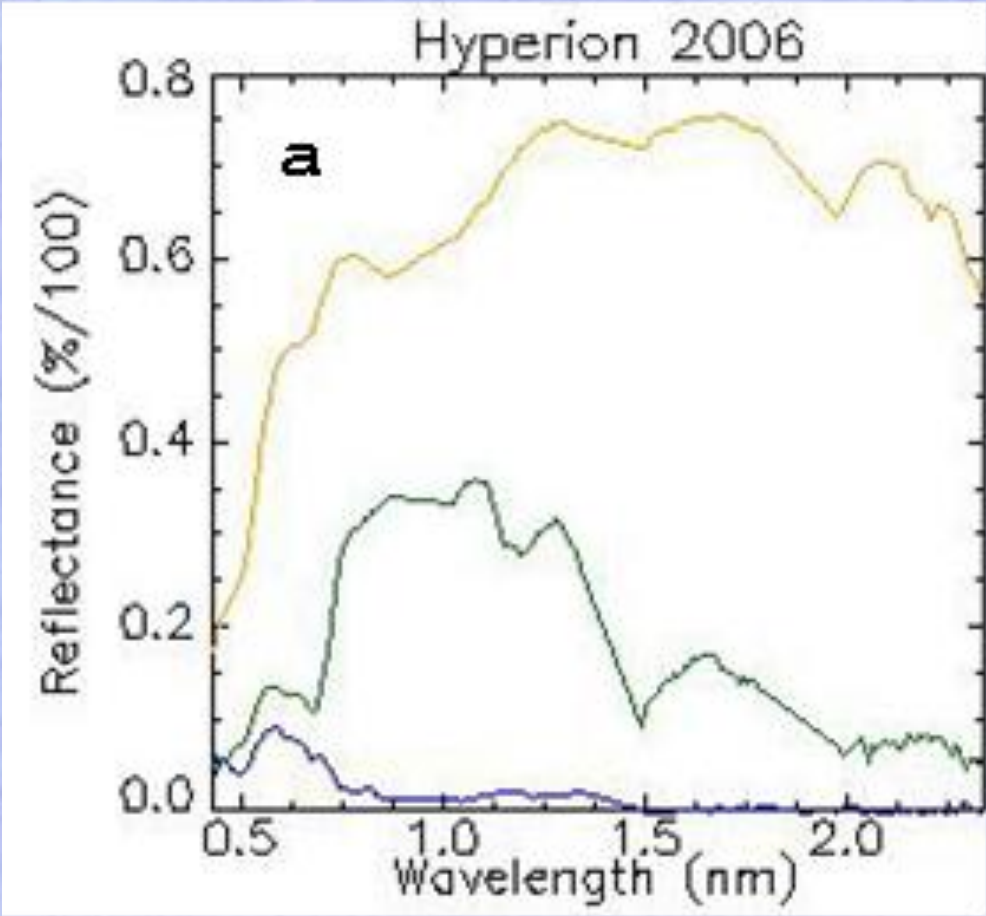
HYMAP



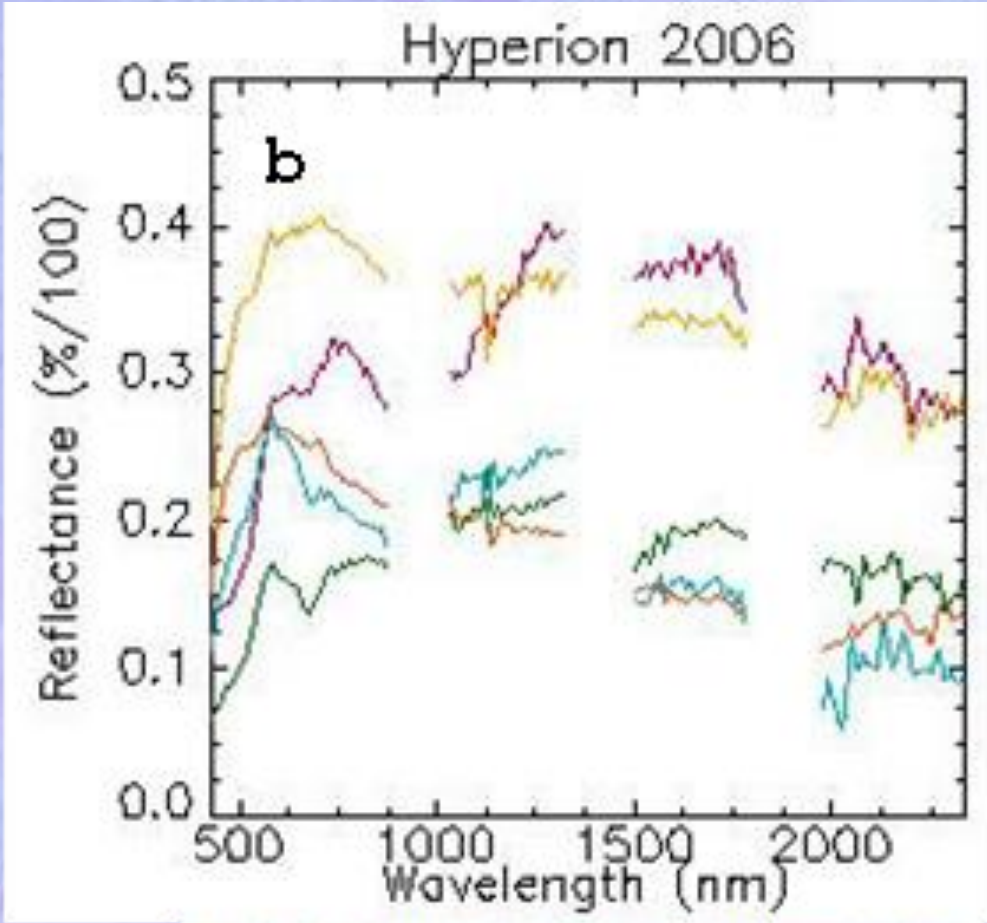
SPECTRA FOR ATMOSPHERIC CORRECTION



SPECTRAL ANALYST WITH REFERENCE SPECTRAL LIBRARIES (Atmospheric correction Empirical Line)

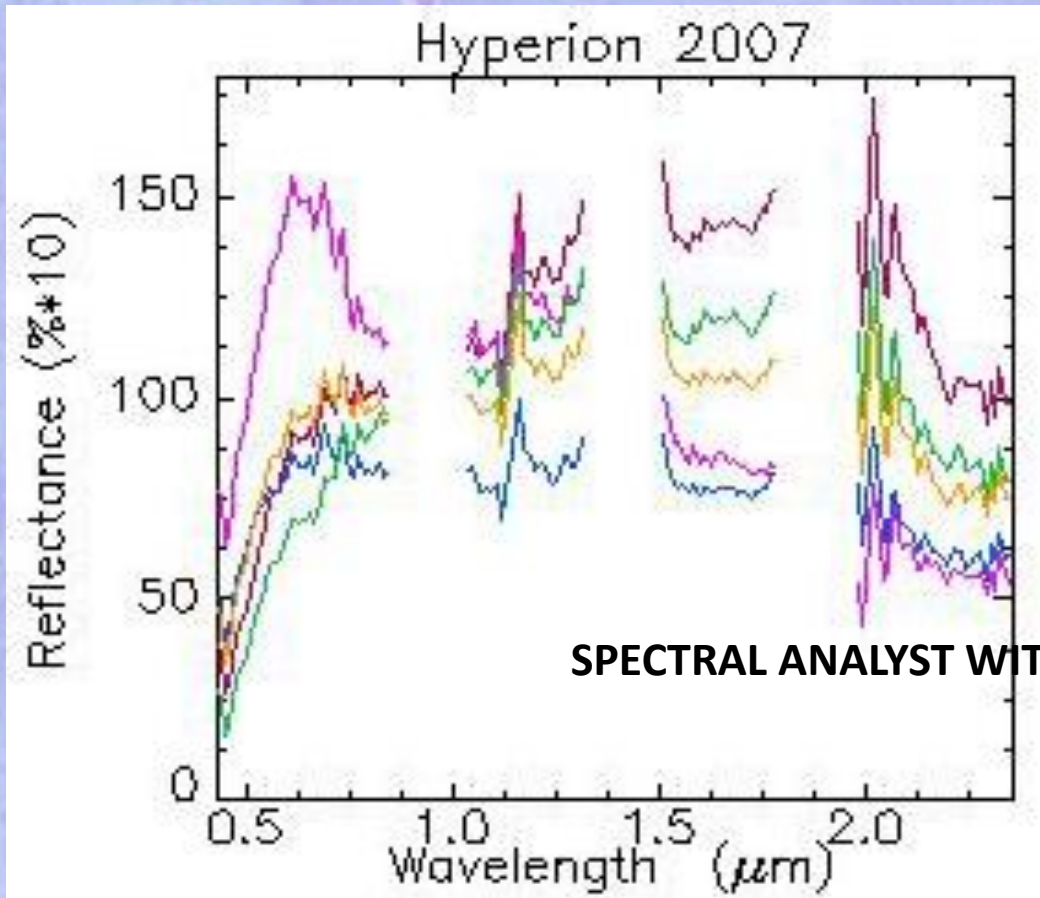


soil
vegetation
acid water

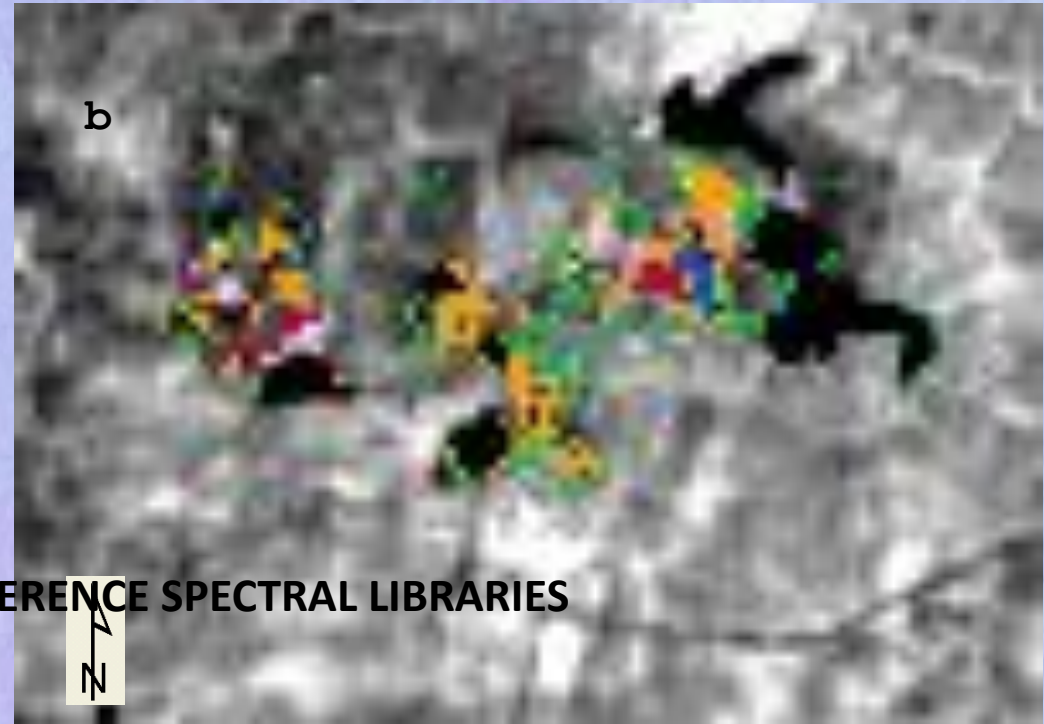


pickeringite holotrichite ferricopiapite
pickeringite holotrichite ferricopiapite
pickeringite holotrichite ferricopiapite
pickeringite holotrichite ferricopiapite
pickeringite holotrichite ferricopiapite

SPECTRAL ANALYST WITH REFERENCE SPECTRAL LIBRARIES (ATCOR4)



SPECTRAL ANALYST WITH REFERENCE SPECTRAL LIBRARIES

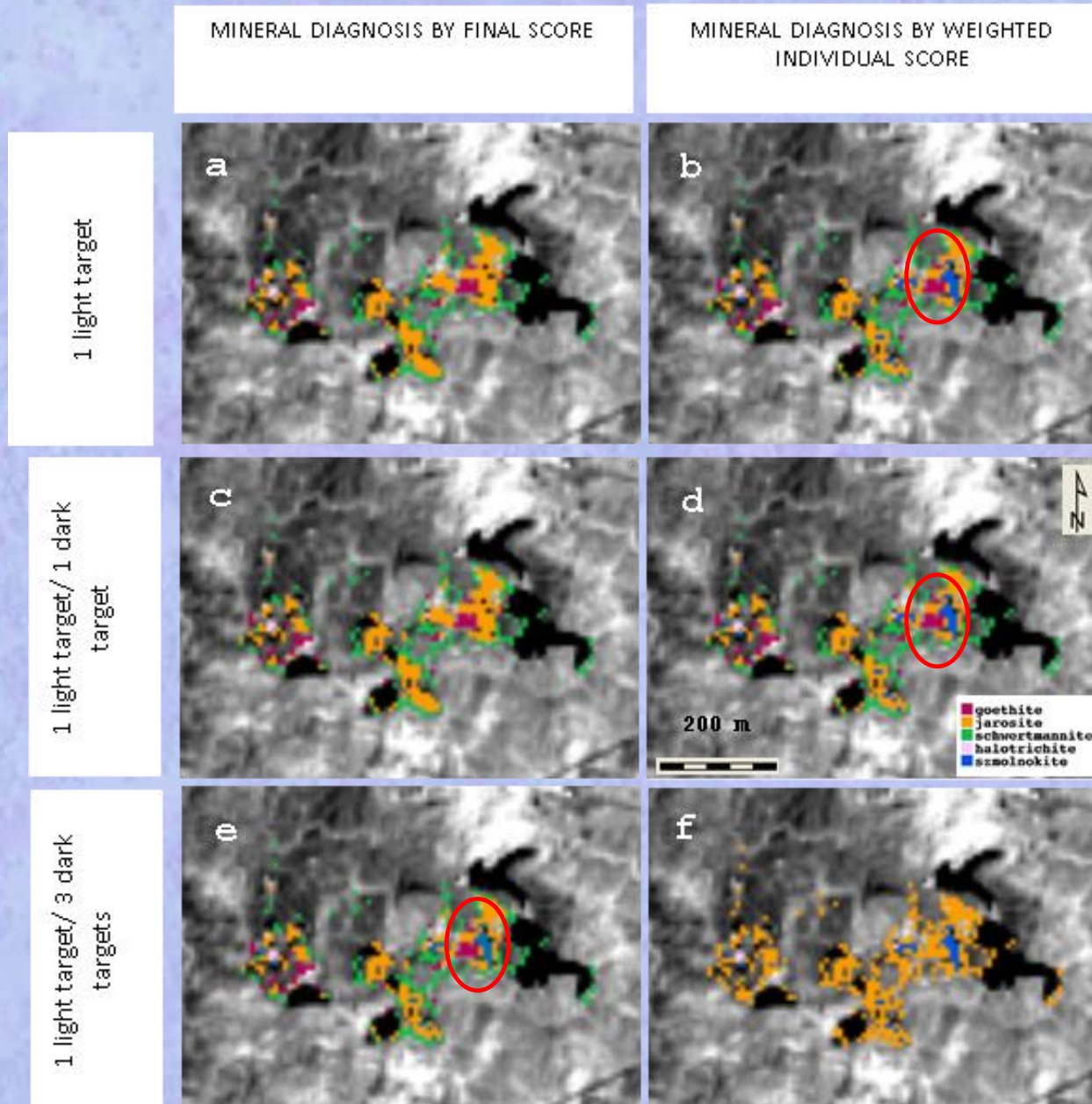


200 m

szmolnokite jarosite rozenite
goethite jarosite schwertmannite
jarosite alunite schwertmannite
halotrichite copiapite rhomboclase
schwertmannite goethite jarosite

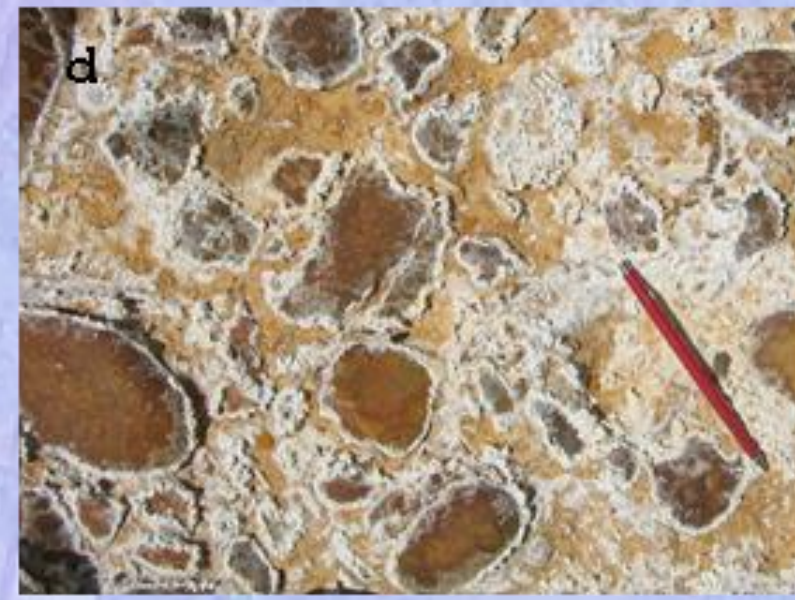
■ goethite
■ jarosite
■ schwertmannite
■ halotrichite
■ szmolnokite

Atmospheric correction



SPECTRAL ANALYST WITH REFERENCE SPECTRAL LIBRARIES (ATCOR4)

HYPERION August 2007



CONCLUSIONES

Image processing should be tailored per scene and per domain within a mine site. Maps which are derived from individual subscenes, and processed using independent processing procedures, broadly agree with each other with respect to changes in oxidation and dehydration mineral phases.

Reference spectral libraries are useful to assess the oxidation or hydration stage of a mineral mixture. In addition, they help establish statistical evaluations of scores produced by mineralogical diagnoses, which are established by the spectral library using various algorithms. Variations in these scores can be used to improve our understanding of contaminant patterns.

The atmospheric correction method has a strong influence on mineralogical diagnoses of spectra derived from hyperspectral imagery. However, calibration targets have to be carefully established to adjust spectral features of the substances to the map, in addition to general landcover features, such as soil, vegetation or water.

Although all geological interpretation is based on a reference spectral library, which only includes a small number of minerals, reality is far more complex. Hence, the maps provided in this study must be understood as simplified images of mineralogical trends in a closed geological context.